ALKALINITY (COLORIMETRIC, AUTOMATED, METHYL ORGANGE) EPA 310.2 1974					
Facility Name:					
Assessor Name:Analyst Name:	Inspection Date				
Relevant Aspect of Standards	Method Reference	Y	N	N/A	Comments
Records Examined: SOP Number/ Revision/ Date Analyst:					nalyst:
Sample ID: Date of Sample Prepar	ation:		Date of Analysis:		
Were samples refrigerated at 4°C?	3.1				
Were sample bottles not opened prior to analysis?	3.1				
Were samples analyzed by this method never filtered?	3.1				
Was methyl orange indicator made by dissolving methyl orange in distilled water at a rate of 0.125 g/L?	6.1				
Was pH 3.1 buffer made by dissolving 5.1047 g potassium acid phthalate and 87.6 mL 0.1N HCl in 1 Liter of distilled water?	6.2				
Was pH 3.1 buffer used for not longer than 1 week?	6.2				
Was methyl orange-buffered indicator made by mixing 1 Liter of pH 3.1 buffer and 200 mL methyl orange solution?	6.3				
Was methyl orange-buffered indicator used for not longer than 24 hours?	6.3				
Was anhydrous sodium carbonate oven-dried at 250°C for 4 hours prior to being used to make standards?	6.4				
Were all reagents run through instrument long enough to obtain stable baselines prior to analysis?	7.2				
Were working standards analyzed in order of decreasing concentration?	7.4				
Were standard curves prepared by plotting the peak heights of standards against their known concentrations?	8.1				
Notes/Comments:					